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best answers each question. \_\_\_\_\_ 1.

### **Assessment Chapter Test B - Wag & Paws**

holt mcdougal modern chemistry 3 chapter test chapter test b, continued 16. the measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. the energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

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for each type of investigation, select the most appropriate branch of chemistry from the following choices: organic chemistry, analytical chemistry, biochemistry, theoretical chemistry. more than one branch may be appropriate. analytical chemistry a. a forensic scientist uses chemistry to find information at the scene of a crime. theoretical ...

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26. according to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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water to produce a sulfuric acid solution that falls to the ground as rain or snow. so  $3(g) + 2o(l) \rightarrow h_2so_4(aq)$  37. have a sour taste; change the color of acid-base indicators; some react with

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section 2 continued 4. a. write the equation for the first ionization of  $\text{H}_2\text{CO}_3$  in aqueous solution. assume that water serves as the reactant that attaches to the hydrogen ion released from the  $\text{H}_2\text{CO}_3$

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